

Figure 1 is a plot of pH versus logtime for the reaction of 10^{-4} M H_2O_2 with 10^{-3} M Fe(II) at 25°C . The plot shows three linear data series with negative slopes, representing different ionic strengths (0.05 M, 0.10 M, and 0.15 M NaClO_4). The y-axis (pH) ranges from 5.25 to 7.25, and the x-axis (logtime) ranges from 0.00 to 2.25. The lines are labeled with their respective ionic strengths: 0.05, 0.10, and 0.15. Data points are marked with '+' symbols.

Plot	+ + +	pH*logtime	_____	PRED*logtime
	_____	L90*logtime	_____	U90*logtime